INAPPLICABILITY OF THE "EXO RULE" TO THE 1,3-DIPOLAR CYCLOADDITIONS TO NORBORNADIENES

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1,3-Dipolar cycloadditions of phenylglyoxylonitrile oxide, benzonitrile-N-phenylimine, or N-phenyl-C-p-nitrophenylnitrone to norbornadiene and 2,3-disubstituted norbornadienes gave the endo-adducts together with the exo-adducts. These observations show that 1,3-dipolar cycloadditions to norbornadienes do not follow the "exo rule" of Alder and Stein.

The "exo rule" was deduced by Alder and Stein from the facts that the 1,3-dipolar cycloadditions of phenyl azide to bicyclic olefins having a moiety of bicyclo-[2.2.1]hydrocarbon skeleton afforded only exo-adducts. Thereafter, many papers have shown that cycloadditions of any 1,3-dipole other than phenyl azide to norbornenes follow the "exo rule". Similarly many 1,3-dipolar cycloadditions to norbornadiene have been reported to follow the "exo rule". For example, the cycloadditions of 1,3-dipoles such as diphenyldiazomethane, dimethyl phenyldiazomethylphosphonate, c,N-diphenylnitrone, N-methyl-C-phenylnitrone, C-propyl-N-cyclohexylnitrone, s,4-dihydroisoquinoline 2-oxide, benzonitrile-N-phenylimine, and formonitrile oxide, to norbornadiene have always afforded only exo-adducts. A survey of the literature, however, showed that a small amount of the endo-adduct had been obtained together with the exo-adduct in the cycloaddition of phenyl azide to norbornadiene, and that the formation of the endo-adduct in the reaction of benzonitrile oxide with norbornadiene had been assumed. Now, one must consider either that the above two cases are special exceptions which do not follow the "exo rule", or that the "exo

rule" is not applicable to 1,3-dipolar cycloadditions to norbornadiene. In order to determine either the former or the latter, we carried out the 1,3-dipolar cycloadditions of phenylglyoxylonitrile oxide($\frac{4}{3}$), benzonitrile-N-phenylimine($\frac{5}{3}$), C,N-diphenylnitrone($\frac{5}{3}$), or N-phenyl-C-p-nitrophenylnitrone($\frac{5}{3}$) to norbornadiene($\frac{1}{3}$), 2,3-bis(methoxycarbonyl)norbornadiene($\frac{2}{3}$) and 2,3-dicyanonorbornadiene($\frac{3}{3}$), and we isolated the endo-adducts in these 1,3-dipolar cycloadditions.

The reaction of 4 with 1 was carried out in the following manner. To a stirred solution of α -chloro- α -hydroxyiminoacetophenone(1.84g, 10.0 mmol) and norbornadiene (2.00g, 21.8 mmol) in THF(30 cm³) in an ice bath, a solution of triethylamine(1.30g, 13.0 mmol) in THF(20 cm³) was added dropwise over one hour. Usual work up of the reaction mixture afforded a mixture of monoadducts and bisadducts(See Table I). The separation of these mono- and bisadducts was carried out by use of column chromatography on silica gel. As Table I shows, other cycloadditions of the 1,3-dipoles (4,5,6a,6b) to the norbornadienes(1,2,3) also gave the endo-adducts together with the exo-adducts.

Each isolated adduct had the appropriate composition(by combustion analysis)

Olefin	1,3- Dipole	Product						
		Monoadduct						Bisadduct
		Yield ^a)) Compo		osition		Exo(%)	Yield ^{a)}
			Exo	Mp(°C)	Endo	Mp(°C)	Endo(%)	(%)
- l	Ą	7,6	දිද (74)	65-66	දිද (26)	53-54	2.8	17
Ę	4	95	දූරු (80)	113-114	<u> გ</u> ხ (20)	151-153	4.0	0
Ą	4	90	දිද (68)	140-141	၃၄ (32)	127-127.5	2.1	0
Ļ	Ę	95	LQ (87)	134-135	貝 (13)	110-111	6.7	4
Į	5	79 ^{b)}	$f\delta_p$)	133-135 ^{c)}	b)			6
Į	6्र	73	12a (44) and	76-77	14년 (0)			7
			and 13a (56)	110-111				
Į.	ጲዼ	54 ^{d)}	13a ^{d)}	111 ^{e)}	d)			2.8
Į	ЯŖ	76		153-154	14b (4)	138-140	24	7
			and 13b (67)	121-122				
Į	Z	67 ^{f)}	إرباً (92)	55 ^{f)}	д́б ^{f)} (8)	124 ^{f)}	11 ^{f)}	g)

Table I: Yield and Product Composition

a) The yields are based on the starting 1,3-dipoles. b) The endo-isomer was not obtained in the reaction reported in Ref.5. c) From Ref.5. d) Only 13a was isolated in the reaction reported in Ref.4. e) From Ref.4. f) From Ref.7. g) A small amount of bisadducts was obtained in the reaction reported in Ref.7.

and IR spectroscopic character. The exo-endo assignments of each isolated adduct were based on the NMR data. In the case of an exo-isomer, the coupling constant between the brigehead proton and the adjacent endo-proton has a value near zero, because the dihedral angle between those protons is nearly a right angle, while the corresponding value for the exo-proton of an endo-isomer is in the range 3-6 Hz $_{ ext{.}}^{9)}$ For example, the two endo-protons(2-H and 6-H) of the exo-adduct(8a) appeared as two doublets(J_{2.6}=9 Hz) at δ 3.78 and 4.99 ppm. The corresponding exo-protons of the endoadduct(9a) appeared as two double doublets($J_{2,6}=9.5$ Hz and $J_{1,2}=J_{6,7}=4$ Hz) at δ 4.09 and 5.37 ppm. The 1,3-dipolar cycloaddition of 4 to 2 or 4 to 3 took place on the more electron-rich double bond to give the adducts (8b and 9b, or 8c and 9c). The NMR data of these adducts are shown below. $\&b(CDCl_3): \delta 1.66$ and 1.90(ABq,2,J=10 Hz,10-CH₂)),3.73(broad,2), 3.78(s,3), 3.81(s,3), 4.05(d,1,J=8.5 Hz,2-endo H), 5.19(d,1,J=8.5 Hz,6-endo H), and 7.3-8.3 ppm(m,5). $9b(CDCl_3): \delta 1.57$ and 1.93(ABq,2,J=10 Hz,10-CH₂), 3.33(s,3), 3.71(m,1), 3.78(s,3), 3.82(m,1), $4.40(dd,1,J=10\ Hz\ and\ J=4\ Hz,2-exo\ H)$, 2.05(ABq,2,J=11.5 Hz,10-CH₂), 3.78(broad,1), 3.92(broad,1), 4.06(d,1,J=9.5 Hz,2-endo H), 5.20(d,1,J=9.5 Hz,6-endo H), and 7.2-8.4 ppm(m,5). 9c (CDCl₃): δ 1.84 and 1.93 $(ABq, 2, J=11 Hz, 10-CH_2)$, 3.96(m,2), 4.55(dd,1,J=10 Hz and J=4 Hz,2-exo H), 5.77(dd,1, J=10 Hz and J=4 Hz, 6-exo H), and 7.5-8.1 ppm(m,5).

The results show that an endo-adduct can be also obtained even in a 1,3-dipolar cycloaddition to norbornadiene which has been reported to give only exo-adduct. From the above results, we conclude that the "exo rule" is not applicable to 1,3-dipolar cycloadditions to norbornadienes.

The reason why 1,3-dipolar cycloadditions to norbornadienes can afford the endo-adducts together with the exo-adducts, and the experimental details of all of the cycloadditions carried out here will be described later elsewhere.

References

- 1) a) K. Alder and G. Stein, Ann. Chem., <u>485</u>, 211, 223(1931).
 - b) K. Alder and G. Stein, ibid., 501, 1(1933).
 - c) K. Alder and G. Stein, ibid., 515, 165, 185(1935).
 - d) K. Alder, H-J. Ache, and F. H. Flock, Chem. Ber., 93, 1888(1960).
 - e) R. Huisgen, R. Grashey, and J. Sauer in S. Patai, "The Chemistry of Alkenes", Intersience, New York 1964, p.739.
- 2) J. W. Wilt and T. P. Malloy, J. Org. Chem., 38, 277(1973).
- 3) H. Cohen and C. Benezra, Can. J. Chem., <u>54</u>, 44(1976).
- 4) R. Huisgen, R. Grashey, H. Hauck, and H. Seidl, Chem. Ber., 101, 2043(1968).
- 5) R. Huisgen, M. Seidel, G. Wallbillich, and H. Knupfer, Tetrahedron, 17, 3(1962).
- 6) R. Huisgen and M. Christl, Angew. Chem. Int. Ed., $\underline{6}$, 456(1967); Chem. Ber., $\underline{106}$, 3291(1973).
- 7) a) S. McLean and D. M. Findlay, Tetrahedron Lett., 2219(1969).
 - b) D. M. Findlay, M. L. Roy, and S. McLean, Can. J. Chem., <u>50</u>, 3186(1972).
- 8) R. Lazar, F. G. Cocu, and N. Barbulescu, Rev. Chim. (Bucharest), <u>20</u>, 3(1969), Chem. Abstr., 70, 114341n(1969).
- 9) a) P. Laszlo and P. von R. Shleyer, J. Am. Chem. Soc., 86, 1171(1964).
 - b) E. I. Synder and B. Franzus, ibid., 86, 1166(1964).
 - c) cf. J. L. Marshall and S. R. Walter, ibid., 96, 6358(1974).

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